Redetermination of 1,7-Diphenyl-1,7dicarba-*closo***-dodecaborane**(12)

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Abstract

The previous determination of the structure of the title molecule, $C_{14}H_{20}B_{10}$ [Astakhin, Romanov, Gusev, Kalinin, Zakharkin & Los (1977). *Zh. Strukt. Khim.* 18, 406–408] with R = 0.15 has been superseded by the present determination which has R = 0.06.

Comment

We are currently investigating the structural and chemical consequences of introducing deliberate overcrowding into heteroborane polyhedra. A substantial number of interesting results have recently been obtained using the $\{7,8-Ph_2-7,8-C_2B_9H_9\}$ ligand, in which the cage-bound phenyl groups are adjacent not only to each other but also to a ligated metal fragment (e.g. Grädler, Reed, Welch & Weller, 1996). In contrast, relatively few studies have so far been reported involving the {7,9-Ph₂-7,9-C₂B₉H₉ ligand. The structure of the parent *closo* carborane, 1,7-Ph₂-1,7-C₂B₁₀H₁₀, has already been reported (Astakhin, Romanov, Gusev, Kalinin, Zakharkin & Los, 1977) but is of limited precision (R = 0.15). To establish an accurate structure of this molecule against which to discuss the structures of future derivatives, we herein report an accurate redetermination of the title compound, (I).



The resulting structure of $1,7-Ph_2-1,7-C_2B_{10}H_{10}$ is accurately defined (e.s.d. on C—C distance typically 0.003 Å). The carborane cage has the expected near-icosahedral geometry with C—B distances in the

© 1996 International Union of Crystallography Printed in Great Britain – all rights reserved range 1.701 (3)–1.732 (3) Å and B—B distances in the range 1.760 (4)–1.787 (4) Å. C_{cage} — C_{phenyl} distances are 1.514 (3) and 1.520 (3) Å. The phenyl substituents are planar within experimental error and stand orthogonal to the B₅ rings to which the cage C atoms are attached [dihedral angles 88.6 (2) and 92.1 (2)°]. The dihedral angle between the two phenyl rings is 113.1 (2)°.

An accurate crystallographic study of $1,2-Ph_2-1,2-C_2B_{10}H_{10}$ (Lewis & Welch, 1993) has been reported recently. A further example of a related molecule is 1-Ph-1,2-C_2B_{10}H_{11} in two crystalline modifications which has recently been structurally characterized (Thomas, Rosair & Welch, 1996; Brain, Cowie, Donohoe, Hnyk, Rankin, Reed, Reid, Robertson, Welch, Hofmann & Schleyer, 1996).



Fig. 1. Perspective view of 1,7-Ph₂-1,7-C₂B₁₀H₁₀ with displacement ellipsoids drawn at the 40% probability level, except for H atoms which have artificial radii of 0.1 Å for clarity.

Experimental

The title compound was synthesized by the thermolysis of $1,2-Ph_2-1,2-C_2B_{10}H_{10}$ in a sealed tube at 670 K for 48 h. The crude product was extracted into pentane and purified by column chromatography using pentane as eluent. Purity was confirmed by microanalysis and ¹¹B NMR spectroscopy. Crystals were grown by the slow evaporation of a pentane solution at 290 K.

Crystal data $C_{14}H_{20}B_{10}$ $M_r = 296.40$

Mo $K\alpha$ radiation $\lambda = 0.71073$ Å

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$C_{14}H_{20}B_{10}$

Monoclinic $P2_1/n$ a = 10.8980 (13) Å b = 7.6993 (6) Å c = 20.713 (2) Å $\beta = 103.971 (8)^{\circ}$ $V = 1686.6 (3) Å^{3}$ Z = 4 $D_x = 1.167 \text{ Mg m}^{-3}$ D_m not measured Data collection	Cell parameters from 29 reflections $\theta = 4.72-12.44^{\circ}$ $\mu = 0.057 \text{ mm}^{-1}$ T = 293 (2) K Block $0.6 \times 0.5 \times 0.4 \text{ mm}$ Colourless
Siemens P4 diffractometer ω scans Absorption correction: empirical via ψ scans (SHELXTL/PC; Sheldrick, 1994) $T_{min} = 0.748$, $T_{max} =$ 0.803 3949 measured reflections 2929 independent reflections 2012 observed reflections $[I > 2\sigma(I)]$	$R_{int} = 0.0371$ $\theta_{max} = 25.00^{\circ}$ $h = -1 \rightarrow 12$ $k = -1 \rightarrow 9$ $l = -24 \rightarrow 24$ 3 standard reflections monitored every 97 reflections intensity decay: 5.9%; corrected for

Refinement

Refinement on F^2	$(\Delta/\sigma)_{\rm max} < 0.001$
$R[F^2 > 2\sigma(F^2)] = 0.0601$	$\Delta \rho_{\rm max} = 0.190 \ {\rm e} \ {\rm \AA}^{-3}$
$wR(F^2) = 0.2465$	$\Delta \rho_{\rm min} = -0.170 \ {\rm e} \ {\rm \AA}^{-3}$
S = 1.020	Extinction correction: none
2927 reflections	Atomic scattering factors
227 parameters	from International Tables
H atoms riding	for Crystallography (1992,
$w = 1/[\sigma^2(F_o^2) + (0.0856P)^2]$	Vol. C, Tables 4.2.6.8 and
+ 0.4235 <i>P</i>]	6.1.1.4)
where $P = (F_0^2 + 2F_c^2)/3$	

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters $(Å^2)$

$U_{eq} = (1$	$(3)\Sigma_i\Sigma_i$	iUija;*	$a_i^* \mathbf{a}_i \cdot \mathbf{a}_j$
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	x	у	z	U_{eq}
Cl	-0.0264(2)	0.4434 (3)	0.15469 (10)	0.0402 (5)
B2	0.0955 (2)	0.2469 (3)	0.13132 (11)	0.0428 (6)
B3	-0.0890 (2)	0.4099 (3)	0.07179 (12)	0.0409 (6)
B4	-0.1033 (3)	0.6141 (4)	0.10786(14)	0.0524 (7)
B5	-0.1195 (3)	0.5758 (4)	0.1905 (2)	0.0573 (8)
B6	-0.1150 (3)	0.3486 (4)	0.20431 (13)	0.0523 (7)
C7	-0.2257 (2)	0.3010(3)	0.06970(10)	0.0419 (5)
B8	-0.2348 (3)	0.5203 (4)	0.0517 (2)	0.0541 (7)
B9	-0.2542 (3)	0.6222 (4)	0.1256(2)	0.0702 (10)
B10	-0.2617 (3)	0.4583 (5)	0.1847 (2)	0.0656 (9)
B11	-0.2469 (3)	0.2540(4)	0.14776 (13)	0.0530 (7)
B12	-0.3325 (3)	0.4246 (4)	0.0991 (2)	0.0598 (8)
C13	-0.2699 (2)	0.1747 (3)	0.01281 (10)	0.0439 (5)
C14	-0.3924 (3)	0.1737 (4)	-0.02450 (13)	0.0675 (8)
C15	-0.4312 (3)	0.0584 (4)	-0.0767 (2)	0.0802 (9)
C16	-0.3487 (3)	-0.0575 (4)	0.09244 (14)	0.0703 (8)
C17	-0.2270 (3)	-0.0580 (4)	-0.05596 (15)	0.0777 (9)
C18	-0.1879 (3)	0.0565 (4)	-0.00398 (14)	0.0677 (8)
C19	0.1167 (2)	0.4520(3)	0.17884 (11)	0.0425 (5)
C20	0.1897 (2)	0.5056 (4)	0.13637 (12)	0.0550 (6)
C21	0.3201 (2)	0.5111 (4)	0.15818 (15)	0.0654 (8)
C22	0.3780 (3)	0.4632 (3)	0.2217 (2)	0.0660 (8)
C23	0.3066 (3)	0.4118 (4)	0.2643 (2)	0.0691 (8)
C24	0.1763 (2)	0.4062(3)	0.24295 (13)	0.0574 (7)

Table 2. Selected geometric parameters (A.	cometric parameters (Å, °)
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C1-C19	1.520 (3)	B5—B9	1.771 (5)
C1—B3	1.707 (3)	B5B6	1.772 (4)
C1-B2	1.707 (3)	B5—B10	1.773 (4)
C1B4	1.725 (3)	B6B10	1.767 (4)
C1-B5	1.727 (3)	B6B11	1.777 (4)
C1-B6	1.732 (3)	C7—C13	1.514(3)
B2—C7	1.715 (3)	C7—B12	1.725 (3)
B2—B6	1.760 (4)	C7—B11	1.726 (3)
B2—B11	1.764 (4)	C7—B8	1.727 (3)
B2—B3	1.773 (3)	B8—B12	1.774 (4)
B3C7	1.701 (3)	B8—B9	1.777 (4)
B3—B8	1.761 (4)	B9—B12	1.765 (4)
B3B4	1.763 (4)	B9B10	1.774 (5)
B4—B8	1.768 (4)	B10B11	1.773 (4)
B4—B9	1.771 (4)	B10-B12	1.773 (4)
B4—B5	1.787 (4)	B11-B12	1.777 (5)
C19-C1-B3	117.8 (2)	С13—С7—В3	116.7 (2)
C19-C1-B2	118.7 (2)	C13—C7—B2	118.3 (2)
C19C1B4	118.2 (2)	C13-C7-B12	120.9 (2)
C19-C1-B5	119.7 (2)	C13-C7-B11	120.3 (2)
C19C1B6	119.9 (2)	С13—С7—В8	118.0(2)

Atomic coordinates for B and C atoms taken from the Cambridge Structural Database (refcode DPMCBO; Allen & Kennard, 1993) were used as a starting point for refinement, after transformation by the matrix (-1,1,0;0,0,-1;0,1,0). The phenyl H atoms were constrained to idealized positions (C-H 0.93 Å) as were the cage H atoms (B-H 1.10 Å). The isotropic displacement parameters of both sets of H atoms were defined as $1.2 \times U_{iso}$ of the bound atom.

Data collection: XSCANS (Siemens, 1994). Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SHELXTL/PC (Sheldrick, 1994). Program(s) used to refine structure: SHELXTL/PC. Molecular graphics: SHELXTL/PC. Software used to prepare material for publication: SHELXTL/PC.

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Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: AB1346). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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